

PHOTOINDUCED PARA-CYCLOADDITION OF PHTHALIMIDES

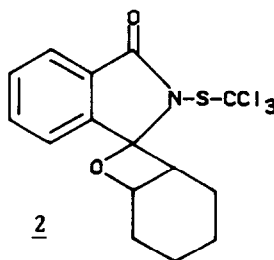
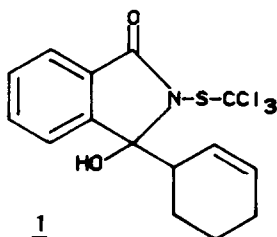
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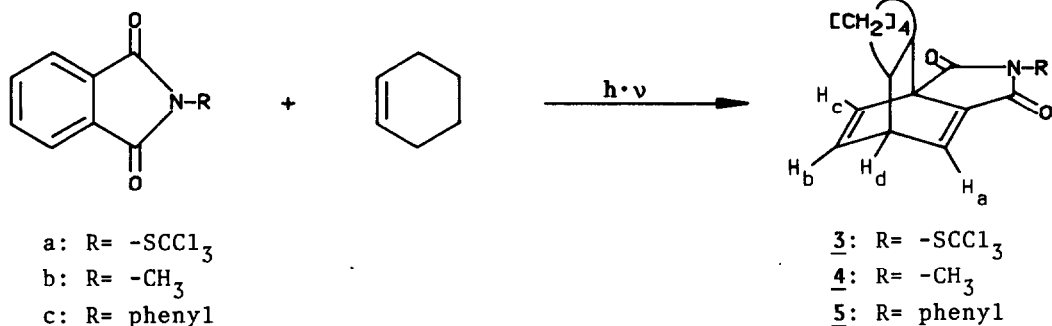
SUMMARY: N-Substituted phthalimides (R= -SCCl₃, -CH₃, phenyl) reacted with cyclohexene on irradiation at $\lambda > 280$ nm by way of para-cycloaddition, yielding N-substituted tricyclo(4.2.2.0)dodeca-8,11-diene-9,10-dicarboximides. The N-(trichloromethylthio)phthalimide gave the cycloadduct as by-product, while for N-methyl- and N-phenylphthalimide the para-cycloaddition predominated.

Many examples of inter- and intramolecular photoadditions of N-substituted phthalimides with alkenes have been published, yielding carbinols (3-alkenyl-3-hydroxyphthalimidines) by allylic attack of the carbonyl group, preferably benzazepinedione derivatives by insertion of the olefin into the C-N bond followed by imide ring enlargement.¹⁾ In this paper a ($4\pi + 2\pi$) para-cycloaddition of the benzene moiety of phthalimides to cyclohexene is described which was observed during photoinduced addition of pesticides to biomolecules²⁾.

During irradiation ($\lambda > 280$ nm)³⁾ of N-(trichloromethylthio)phthalimide, used as a fungicide in plant protection, it is mainly added to cyclohexene by allylic attack of the excited carbonyl group, yielding the corresponding carbinol 1⁴⁾ (80%). The expected oxetane 2⁵⁾ was observed as by-product (6%).⁶⁾



As second but unexpected by-product, 3 (11%⁶⁾), was formed and identified on the basis of its spectra⁸⁾. Instead of the phthalimide AA'BB' pattern in the aromatic region there are signals of three olefinic protons in the ¹H-NMR spectrum, i.e. δ 7.2 (d, J=6 Hz, H_a), 6.82 (dd, J=7 Hz, J=6 Hz, H_b) and 6.15 (dt, J=7 Hz, J=1 Hz, H_c). From decoupling experiments, two of these are vicinal (H_b and H_c) and the third (H_a) is separated by H_d at δ 3.81 (t, J=6 Hz).

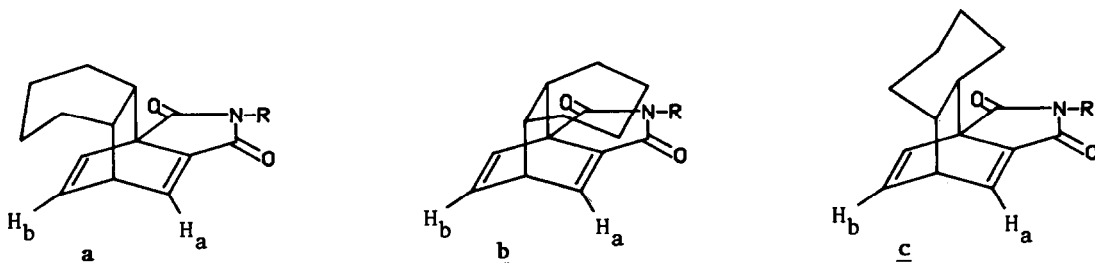


From the three possible 1.2-, 1.3-, 1.4-additions⁷⁾, only the 1.4-(para)-adduct 3 is consistent with the observed pattern of three olefinic protons, providing support for the structure of 3. On GC and HPLC analyses 3 eluted as a sharp peak, establishing its isomeric purity.

Irradiation of N-methylphthalimide in the presence of cyclohexene under the same conditions afforded the para-cycloadduct 4 as main product (44%⁶⁾), but as a mixture of three isomers as was evident from the GC and HPLC analyses.⁹⁾ In the ¹H-NMR spectrum of the isomer mixture of 4, two of the isomers gave the same characteristic olefin signals¹⁰⁾ as 3. The third isomer showed an ABX-pattern for the H_c, H_b, and H_d protons and a doublet for H_a¹⁰⁾. By means of integration the isomer ratio was found to be 10:8:2.

As third example of this type of cycloaddition, the N-phenylphthalimide was irradiated in the presence of cyclohexene, yielding the para-cycloadduct 5¹¹⁾ as main product (>90%⁶⁾). 5 was also obtained in form of three isomers in nearly the same ratio as 4.

The differences between the observed isomers a, b, and c of 4 and 5 may be explained by the configuration of the bridging cyclohexane ring, taking the cis- (isomer a and b) or trans- (isomer c) arrangements. In the cis-isomer a the proton H_b should be shielded downfield due to steric effects. The same can be expected for H_a in the isomer b, while the trans-configuration (isomer c) leaves nearly the same effects on both sides. These estimated effects are consistent with the observed resonances in the ¹H-NMR spectra of 4 and 5, suggesting that the preferred isomer of 3 has the stereochemistry of structure a.



The reason for the stereoselective cycloaddition of cyclohexene to N-(trichloromethylthio)phthalimide is not clear, but appears to depend on the trichloromethylthio substituent.

The ^{13}C -NMR spectra of the isomer mixtures of 4 and 5 exhibit the same characteristic signals in the olefinic region as found for 3⁸⁾. However, these spectra can only be helpful for the stereochemical discussions once the pure isomers have been isolated.

The N-substituted tricyclo(4.2.2.0)dodeca-8,11-diene-9,10-dicarboximides 3, 4 and 5 were all obtained as colourless oils, which unfortunately precluded X-ray analyses of these cycloadducts; but their spectral and analytical data reported here leave no doubt about the assigned structures.

The author thanks Prof. W. Adam for helpful discussions.

REFERENCES AND NOTES

- For recent reviews see: a) Kanaoka, Y., Acc. Chem. Res. **11**, 407 (1978); b) Mazzocchi, P.H., "Organic Photochemistry", vol. 5; Pawda, A. (editor), Marcel Dekker, New York, 1981, p. 421; c) Maruyama, K., Kubo, Y., J. Org. Chem. **50**, 1426 (1985).
- Schwack, W., Habilitationschrift, Würzburg (1986).
- In a general procedure, 100 mg of the N-substituted phthalimide, dissolved in 50 ml cyclohexene (dist. over P_2O_5), was irradiated in a quartz tube for 10 h, using a 150 W high pressure mercury lamp. A glass filter (WG 295, Schott) prevented irradiation with wavelengths shorter than 280 nm. For product isolation column chromatography (SiO_2) with benzene as eluant was used. Products were further purified by preparative TLC (SiO_2 , 5% ethanol in n-hexane).
- Compound 1. m.p. 170-172°C (benzene). IR (KBr): 3440, 3080, 2940, 2870, 1715 (C=O), 1615, 1600, 1360, 1225, 1080, 800 cm^{-1} . MS (15 eV): m/z = 378 (Cl_3 , 5 %, $\text{M}^+ + 1$), 296 (Cl_3 , 29 %), 260 (Cl_2 , 100 %), 243 (36 %), 210 (23 %), 194 (14 %), 130 (48 %), 81 (21 %). $^1\text{H-NMR}$ (400 MHz, CDCl_3): δ 7.91 (1H, dt, J=7.5 Hz, J=1 Hz); 7.61-7.66 (2H, 2 td, J=7.5 Hz); 7.54-7.59 (1H, 2 dd, J=7.5 Hz, J=2.2 Hz); 6.20 (1H, d, J=10 Hz); 5.99 (1H, dq, J=10 Hz, J=2.5 Hz); 3.26 (1H, m); 3.06 (1H, s, OH). $\text{C}_{15}\text{H}_{14}\text{Cl}_3\text{NO}_2\text{S}$ (378.71), calc. % (found %): C 47.57 (47.26), H 3.73 (3.59), N 3.70 (4.06).
- Compound 2. m.p. 135-136°C (ether). IR (KBr): 3080, 2950, 2870, 1740 (C=O), 1610, 1465, 1285, 1030, 960 (oxetane), 800 cm^{-1} . MS (15 eV): m/z = 377 (Cl_3 , 2 %, M^+), 306 (Cl_3 , 4 %), 296 (Cl_3 , 12 %), 260 (Cl_2 , 100 %), 232 (80 %), 82 (95 %), 67 (30 %). $^1\text{H-NMR}$ (400 MHz, CDCl_3): δ 7.87-7.92 (2H, 2 dt, J=7.5 Hz, J=1 Hz); 7.77 (1H, td, J=7.5 Hz, J=1 Hz); 7.57 (1H, td, J=7.5 Hz, J=1 Hz); oxetane protons: 5.15 (1H, dt, J=7.8 Hz, J=6.4 Hz); 3.45 (1H, dt, J=7.8 Hz, J=9.6 Hz). $\text{C}_{15}\text{H}_{14}\text{Cl}_3\text{NO}_2\text{S}$ (378.71), calc. % (found %): C 47.57 (47.36), H 3.73 (3.60), N 3.70 (3.92).

- 6) Yields were calculated on the amount of converted starting material.
- 7) For recent reviews see: Bryce-Smith, D., Gilbert, A., Tetrahedron **32**, 1309 (1976) (Part I); Tetrahedron **33**, 2459 (1977) (Part II).
- 8) Compound 3. IR (film): 3070, 2990, 2950, 2865, 1780, 1730 (C=O), 1660, 1450, 1270, 1250, 1195, 1030 cm^{-1} . MS (15 eV): $m/z = 377$ (Cl_3 , M^+ , 1 %), 342 (Cl_2 , 21 %), 309 (Cl_3 , 23 %), 295 (Cl_3 , 5 %), 260 (Cl_2 , 76 %), 82 (100 %), 67 (59 %). $\text{C}_{15}\text{H}_{14}\text{Cl}_3\text{NO}_2\text{S}$ (378.71), calc. % (found %): C 47.57 (48.06), H 3.73 (3.83), N 3.70 (3.58). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3): δ 27.4, 27.6, 30.2, 32.7 ($-\text{CH}_2-$); 45.8, 50.6, 52.9 ($-\text{CH}<$); 56.4 ($>\text{C}<$); 123.4, 138.3, 141.4 ($=\text{CH}-$); 142.7 ($=\text{C}<$); 162.1, 173.1 ($>\text{C}=\text{O}$). UV (ethanol): λ_{max} 240 nm ($\log \epsilon$ 3.92).
- 9) Kanaoka and Hatanaka (Chem. Pharm. Bull. **22**, 2205 (1974)) irradiated II in the presence of cyclohexene, but isolated only the corresponding carbinol in poor yield by preparative TLC.
- 10) Compound 4. IR (film): 3070, 2940, 2860, 1770, 1705, 1420, 1370, 1255, 995 cm^{-1} . MS (70 eV): $m/z = 243$ (M^+ , 3 %), 175 (46 %), 174 (37 %), 162 (66 %), 161 (34 %), 115 (12 %), 82 (29 %), 81 (10 %), 67 (100 %), 54 (66 %), 41 (46 %). $^1\text{H-NMR}$ (CDCl_3): 4a (main isomer): δ 6.92 (1H, d, $J=6$ Hz, H_a); 6.77 (1H, dd, $J=7$ Hz, $J=6$ Hz, H_b); 6.07 (1H, dt, $J=7$ Hz, $J=1$ Hz, H_c); 3.65 (1H, t, $J=6$ Hz, H_d). 4b: δ 7.51 (1H, d, $J=6$ Hz, H_a); 6.80 (1H, dd, $J=7$ Hz, $J=1.5$ Hz, H_c); 6.08 (1H, dd, $J=7$ Hz, $J=6$ Hz, H_b); 3.67 (1H, t, $J=6$ Hz, H_d). 4c: δ 7.16 (1H, dd, $J=6$ Hz, $J=0.8$ Hz, H_a); 6.38-6.45 (2H, AB-part of ABX, H_bH_c); 3.77 (1H, tt, X-part of ABX, H_d).
- 11) Compound 5. IR (film): 3070, 3040, 2930, 2860, 1770, 1710, 1600, 1500, 1370, 1190 cm^{-1} . MS (70 eV): $m/z = 305$ (M^+ , 9 %), 237 (62 %), 236 (69 %), 223 (100 %), 222 (52 %), 179 (52 %), 178 (31 %), 115 (14 %), 82 (7 %), 77 (31 %), 76 (38 %), 67 (48 %), 54 (38 %), 41 (66 %). $^1\text{H-NMR}$ (CDCl_3): 5a (main-isomer): δ 7.06 (1H, d, $J=6$ Hz, H_a); 6.83 (1H, dd, $J=7$ Hz, $J=6$ Hz, H_b); 6.18 (1H, dt, $J=7$ Hz, $J=1$ Hz, H_c); 3.73 (1H, t, $J=6$ Hz, H_d). 5b: δ 7.65 (1H, d, $J=6$ Hz, H_a); 6.91 (1H, dd, $J=7$ Hz, $J=1.5$ Hz, H_c); 6.13 (1H, dd, $J=7$ Hz, $J=6$ Hz, H_b); 3.74 (1H, t, $J=6$ Hz, H_d). 5c: δ 6.41-6.52 (2H, AB-part of ABX, H_bH_c); 3.86 (1H, X-part of ABX, H_d); H_a -signal obscured by the N-phenyl aromatic protons.

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